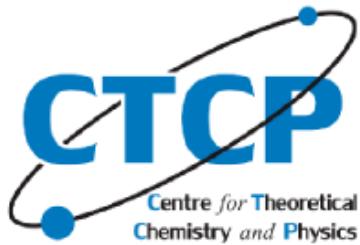


# Benchmark Calculations of Atomic Properties of Elements 112 to 122

A. Borschevsky, V. Pershina, E. Eliav, and U. Kaldor



# Properties under study:

- Ionization potential
- Electron affinity
- Polarizability

# Motivation:

- Fundamental knowledge: relativistic effects, structure of the periodic table, etc.
- Link to chemical/physical properties: reactivity, electronegativity, adsorption enthalpy...

# Methods

- Relativity
- Electron correlation
- Basis sets

# Relativity

## Classical example: gold atom (eV)

			NR	DC	Expt.
IP	$5d^{10}6s$	$^2S_{1/2}$	7.06	9.21	9.23
EE	$5d^96s^2$	$^2D_{3/2}$	5.30	2.66	2.66
	$5d^96s^2$	$^2D_{5/2}$	5.30	1.12	1.14
	$5d^{10}6p_{1/2}$	$^2P_{1/2}$	3.31	4.72	4.63
	$5d^{10}6p_{3/2}$	$^2P_{3/2}$	3.31	5.18	5.11
EA	$5d^{10}6s^2$	$^1S_0$	1.28	2.29	2.31

- strong stabilization and contraction of the  $s$  orbitals
- stabilization of the  $p_{1/2}$  orbitals
- destabilization and expansion of the  $p_{3/2}$ ,  $d$  orbitals

## For SHE:

### 1. $^2S \rightarrow ^2D_{5/2}$ transition (eV)<sup>1</sup>

	Cu	Ag	Au	E111
NR	1.39	3.75	5.30	5.43
DCB	1.39	3.75	1.15	-2.95



### 2. Electron affinity of 118<sup>2</sup>

Predicted electron affinity of element 118 is **0.056** eV, NR calculations give no electron affinity

Ground state of element 111 is  **$6d^97s^2\ ^2D_{5/2}$**   
NR calculations yield  **$6d^{10}7s\ ^2S$** , same as the other coinage metals

<sup>1</sup>Eliav et al, Phys. Rev. Lett. **73**, 3203 (1994)

<sup>2</sup>I. Goidenko, PRA **67**, 020102 (2003)

# Relativity

Classical example: gold atom (eV)

			NR	DC	Expt.
IP	$5d^{10}6s$	$^2S_{1/2}$	7.06	9.21	9.23
EE	$5d^96s^2$	$^2D_{3/2}$	5.30	2.66	2.66
	$5d^96s^2$	$^2D$	5.30	1.12	1.14

- strong stabilization and contraction of the  $s$  orbitals
- stabilization of the  $p_{1/2}$  orbitals
- destabilization and expansion of the

Any investigation of the structure, spectroscopy and chemistry of SHE must include relativity from the start; treating it as a perturbation is inadequate

	Cu	Ag	Au	111
NR	1.39	3.75	5.30	5.43
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# Relativity

- 4-c Dirac-Coulomb-Breit Hamiltonian:

$$H_{DCB} = \sum_i h_D(i) + \sum_{i < j} (1/r_{ij} + B_{ij}),$$
$$h_D(i) = c\alpha_i \cdot \mathbf{p}_i + c^2 \beta_i + V_{nuc}(i)$$

- $\alpha_i, \beta_i$ - four dimensional Dirac matrices
- $V_{nuc}$ - nuclear attraction operator (finite nucleus model)

- $B_{ij} = -\frac{1}{2r_{ij}} [\alpha_i \cdot \alpha_j + (\alpha_i \cdot \mathbf{r}_{ij})(\alpha_j \cdot \mathbf{r}_{ij})/r_{ij}^2]$ , Breit term

correct up to the order of  $\alpha^2$ , higher order corrections (QED) are possible

- 4 component wave functions:  $\Psi = \begin{pmatrix} \Psi^L \\ \Psi^S \end{pmatrix}$

# Electron correlation

Correlation (explicit electronic interaction) is important for both light and heavy systems, and for all properties

- Ionization potentials

	DF	MP2	CCSD	CCSD(T)	Exp.
I	9.663	10.028	10.362	10.436	10.451
At	8.437	9.213	9.241	9.317	-
117	6.619	7.692	7.554	7.650	-

- Electron affinities

Element 118: without correlation: EA<0

with correlation EA=0.056 eV

- Polarizabilities

	DF	MP2	CCSD	CCSD(T)	Exp.
Hg	44.90	27.47	35.31	34.14	33.9 (0.3)
112	29.46	25.11	27.66	27.64	-

# Correlation methods

## Some correlation methods:

- Configuration interaction (CI): Full CI/CISD
- Multiconfigurational Dirac-Fock (MCDF)
- Perturbation methods (MP2, CASPT2)
- Coupled cluster (CC) method

Most powerful (yet most computationally expensive) approach.

$$\Psi = \exp(S)\Psi_0 = \left(1 + S + \frac{S^2}{2!} + \dots\right) \Psi_0$$

$S$  is the excitation operator (usually truncated at  $S_2$ ):

$$S = S_1 + S_2 + \dots + S_N; \quad S_1 = \sum_{ia} s_i^a a_a^\dagger a_i; \quad S_2 = \sum_{ijab} s_{ij}^{ab} a_a^\dagger a_b^\dagger a_j a_i$$

CC equations:

$$\text{CCSD: } \Psi_{\text{CCSD}} = \exp(S_1 + S_2) |\Phi_0\rangle.$$

$$\langle \Phi_0 | (H - E_{\text{CCSD}}) \exp(S_1 + S_2) | \Phi_0 \rangle = 0$$

# Correlation methods

## Some correlation methods:

- Configuration interaction (CI): Full CI/CISD
- Multiconfigurational Dirac-Fock (MCDF)
- Perturbation methods (MP2, CASPT2)
- Coupled cluster (CC) method

- Highly accurate
- Equivalent to infinite order perturbation expansion
- Size extensive (energy scales linearly with num. of electrons)
- Fulfils the separability condition

$$S = S_1 + S_2 + \dots + S_N; \quad S_1 = \sum_{ia} s_i^a a_a^\dagger a_i; \quad S_2 = \sum_{ijab} s_{ij}^{ab} a_a^\dagger a_b^\dagger a_j a_i$$

# Coupled cluster methods

**CCSD(T)** - relativistic single reference coupled cluster

Closed shell systems/systems with one dominant configuration

**Open shell systems, transition energies, bond dissociation:  
multireference coupled cluster**

**FSCCSD** - Fock space coupled cluster

- open shell systems
- many transition energies in a single calculation
- very high accuracy
- limitations:
  1. two valence holes/two valence particles only:  
**application to systems with more valence particles is in development, E. Eliav, Tel Aviv group**
  2. convergence difficulties limit the size of the mode space:  
**Intermediate Hamiltonian (IH) approach**

# Intermediate Hamiltonian approach (IH)

Developed in the Tel Aviv group, three variants: **IH1**, **MSIH** and **XIH**.

Allows for use of extremely large model spaces, while avoiding convergence difficulties. **Benchmark accuracy often requires very large model spaces.**

## Ionization potential of Ra with XIHFCC (exp. IP=5.278 eV)

Model space	IP (eV)	Error (eV)
8s7p6d	5.380	0.102
8s9s7p8p6d7d5f	5.332	0.054
8-11s7-10p6-9d5-6f5g	5.313	0.035
8-14s7-11p6-10d5-7f5-6g	5.300	0.022
8-15s7-12p6-10d5-8f5-6g4h	5.284	0.006
8-15s7-12p6-11d5-9f5-6g4h	5.282	0.0037

Same accuracy  
can be expected  
for the obtained  
IP of element  
**120 (5.888 eV)**

# Basis sets

- **Faegri dual family basis sets**

K. Faegri, Theor. Chem. Acc. **105** 252 (2001)

- **Universal basis set of Malli**

Malli *et al*, PRA **47**, 143 (1993)

# Software

- Tel Aviv atomic computational package

E. Eliav, U. Kaldor, and Y. Ishikawa

- DIRAC08/DIRAC10 computational package

DIRAC, a relativistic ab initio electronic structure program, release  
DIRAC08 (2008)/release DIRAC10 (2010)

# Basis sets

- **Faegri dual family basis sets**

K. Faegri, Theor. Chem. Acc. **10**

- **Universal basis set of M**

Malli *et al*, PRA **47**, 143 (1993)

- Augmented by high  $l$  functions (up to  $l=6$ )
- Extended to convergence

# Software

- Tel Aviv atomic computational package

E. Eliav, U. Kaldor, and Y. Ishikawa

- DIRAC08/DIRAC10 computational package

DIRAC, a relativistic ab initio electronic structure program, release

DIRAC08 (2008)/release DIRAC10 (2010)

# Atoms of interest:

1A		Periodic Table of Elements																		8A	
1 H 1.00794	2A Li 6.941 3 Be 9.012182											2 He 4.002602									
3 Li 6.941 11 Na 22.989769	4 Be 9.012182 12 Mg 24.3050											2 He 4.002602									
19 K 39.0983	20 Ca 40.078 21 Sc 44.955912	22 Ti 47.987 23 V 50.9415 24 Cr 51.9961 25 Mn 54.938045 26 Fe 55.845 27 Co 58.933195 28 Ni 58.6934 29 Cu 63.546 30 Zn 65.38 31 Ga 69.723 32 Ge 72.64 33 As 74.92160 34 Se 78.96 35 Br 79.904 36 Kr 83.798	3B 3 B 10.811 4B 4 C 12.0107 5B 5 N 14.0067 6B 6 O 15.9994 7B 7 F 18.9984032 8B 8 Ne 20.1797	1B 1 Al 13 14 Si 30.973762 2B 2 P 32.065 3 S 35.453 7A 7 Cl 39.948 8A 8 Ar 39.948																	
37 Rb 80.983 38 Sr 87.675 39 Y 88.90184	40 Zr 91.224 41 Nb 92.90638 42 Mo 95.96 43 Tc [98] 44 Fe 55.845 45 Co 58.933195 46 Ni 58.6934 47 Cu 63.546 48 Zn 65.38 49 Ga 69.723 50 Ge 72.64 51 As 74.92160 52 Se 78.96 53 Br 79.904 54 Kr 83.798	80 Hg 81 Tl 82 Pb 83 Bi 84 Po 85 At 86 Rn																			
87 Fr 223.01407 88 Ra 226.02570	72 Hf 178.49 73 Ta 180.94788 74 W 183.84 75 Re 186.207 104 Rf [267] 105 Db [268] 106 Sg [271] 107 Bh [272]	112 Cn 113 Uut 114 Uuq 115 Uup 116 Uuh 117 Uus 118 Uuo																			
119 Nh 284.715 120 Fl 285.050	57 Fr 223.01407 58 Ra 226.02570 59 Ac 227.02891 60 Th 229.02891 61 Pa 231.02891 62 U 232.02891 63 Np 237.02891 64 Pu 244.02891 65 Am 243.02891 66 Cm 247.02891 67 Bk 247.02891 68 Cf 251.02891 69 Es 252.02891 70 Fm 257.02891 71 Md 258.02891 72 No 259.02891 73 Lr 262.02891	142 Nh 223.01407 143 Ra 226.02570 144 Ac 227.02891 145 Th 229.02891 146 Pa 231.02891 147 U 232.02891 148 Np 237.02891 149 Pu 244.02891 150 Am 243.02891 151 Cm 247.02891 152 Bk 247.02891 153 Cf 251.02891 154 Es 252.02891 155 Fm 257.02891 156 Md 258.02891 157 No 259.02891 158 Lr 262.02891																			
Actinides		121 Nh 284.715 122 Ra 226.02570	121 Nh 284.715 122 Ra 226.02570																		

# Results

- Ionization potentials
- Electron affinities
- Polarizabilities

# Ionization potentials – 7p elements

	Hg	Tl	Pb	Bi	Po	At	Rn
IHFSCC	10.455 <sup>1</sup>	6.096	7.349 <sup>2</sup>	7.304	8.499	9.317	10.799
Other calc.	10.39 <sup>3</sup> ECP-CCSD(T)		7.044 <sup>4</sup> MCDF			9.35 <sup>5</sup> MCDF	10.482 <sup>3</sup> ECP-CCSD(T)
IP exp. (eV)	10.438	6.108	7.417	7.286	8.414	-	10.749
	Cn	113	114	115	116	117	118
IHFSCC (eV)	11.970 <sup>1</sup>	7.420	8.539 <sup>2</sup>	5.553	6.881	7.650	8.914
Other calc.	11.68 <sup>3</sup> ECP-CCSD(T)	7.4 <sup>7</sup> DF	8.286 <sup>4</sup> MCDF	5.5 <sup>6</sup> DF	7.5 <sup>6</sup> DF	7.64 <sup>5</sup> MCDF	8.6423 <sup>3</sup> ECP-CCSD(T)

<sup>1</sup> E. Eliav *et al.*, PRA **52**, 2765 (1995)

<sup>2</sup> A. Landau *et al.*, JCP **114**, 2977 (2001)

<sup>3</sup> C. Nash, JPC A **109**, 3493 (2005)

<sup>4</sup> Y.J. Yu *et al.*, JCP **128**, 124316 (2008)

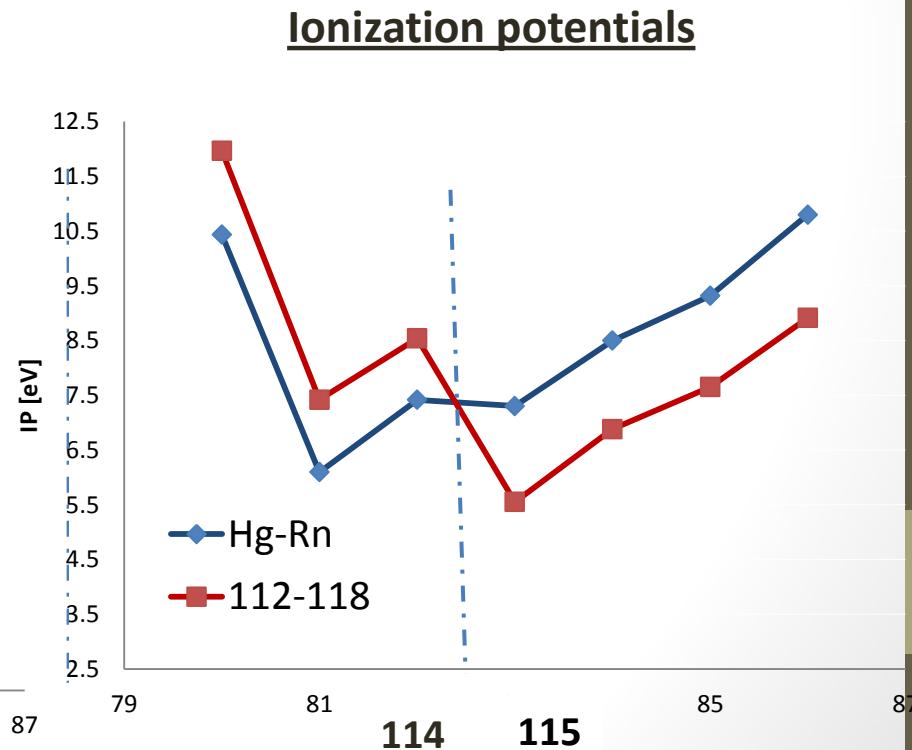
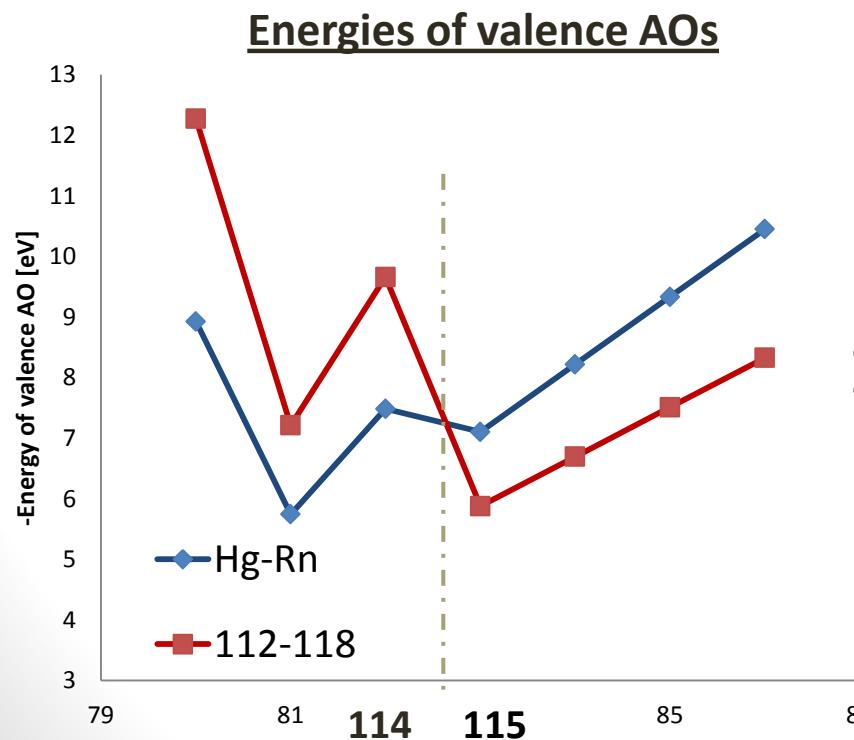
<sup>5</sup> Z. Chang *et al.*, JPC A **114**, 13388 (2010)

<sup>6</sup> B. Fricke, Struct. Bonding **21**, 89 (1975)

<sup>7</sup> Keller *et al.*, JPC **74**, 1127 (1970)

# Ionization potentials – 7p elements

	Hg	Tl	Pb	Bi	Po	At	Rn
IHFSCC	10.455 <sup>1</sup>	6.096	7.349 <sup>2</sup>	7.304	8.499	9.317	10.799
	Cn	<b>113</b>	<b>114</b>	<b>115</b>	<b>116</b>	<b>117</b>	<b>118</b>
IHFSCC (eV)	11.970 <sup>1</sup>	7.306	8.539 <sup>2</sup>	5.553	6.881	7.650	8.914



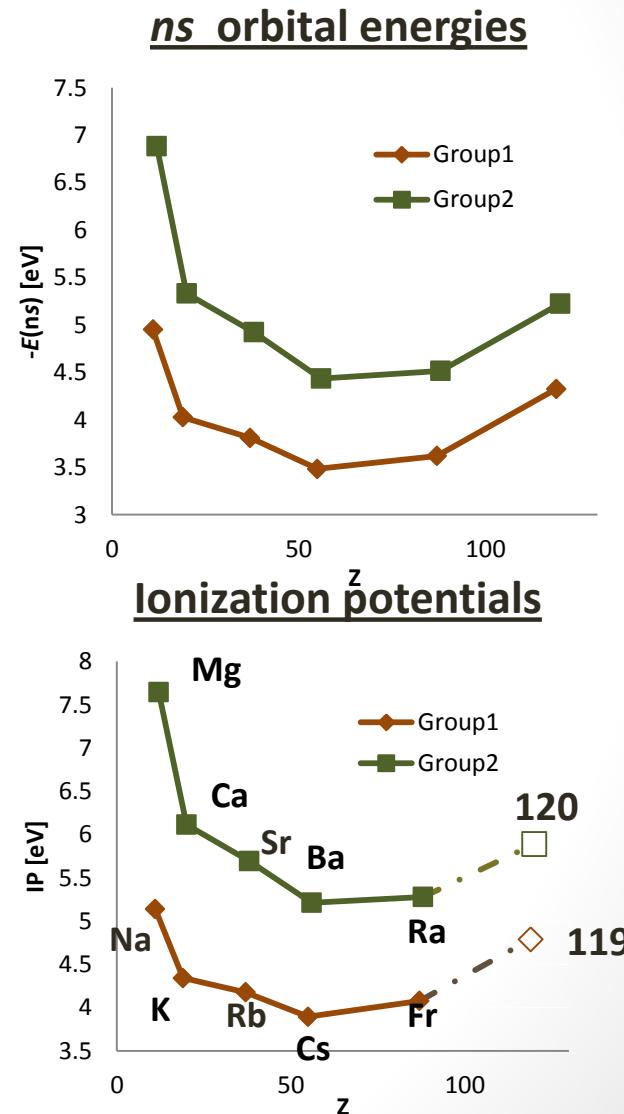
# Ionization potentials – 8s elements

	Fr	Ra
IHFSCC	4.077 <sup>1</sup>	5.282
DK+CCSD(T)	4.038 <sup>2</sup>	
IP exp. (eV)	4.073	5.278
	<b>119</b>	<b>120</b>
IP calc (eV)	4.788 <sup>1</sup>	5.888
DK+CCSD(T)	4.713 <sup>2</sup>	6.0 ( <b>DF</b> ) <sup>3</sup>

<sup>1</sup> E. Eliav *et al.*, Chem. Phys. **311**, 163 (2005)

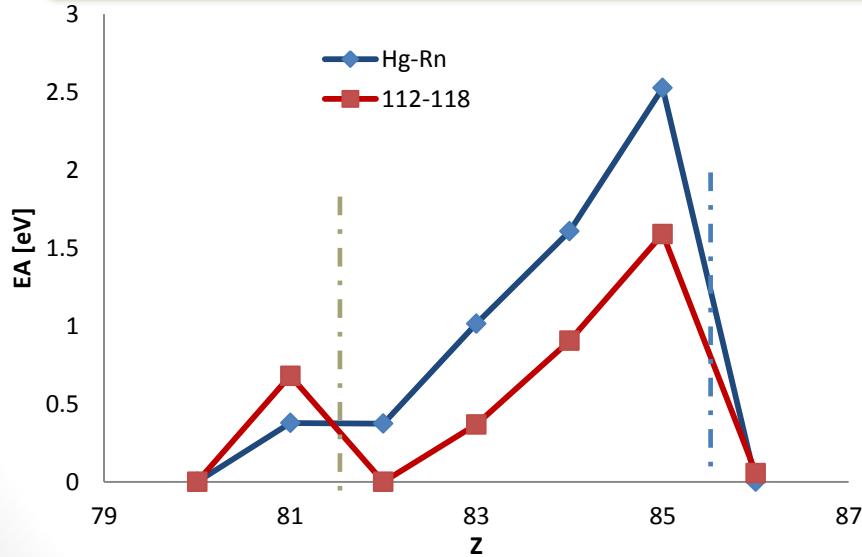
<sup>2</sup> I. Lim *et al.*, JCP **122**, 104103 (2005)

<sup>3</sup> B. Fricke, Struct. Bonding **21**, 89 (1975)



# Electron affinities – 7p and 8s elements

	Hg	Tl	Pb	Bi	Po	At	Rn
IHFSCC (eV)	-	0.410 <sup>1</sup>	0.373	1.015	1.608	2.527	-
EA exp. (eV)	-	0.377	0.364	0.942	1.9 (3)	2.38 <sup>2</sup> ( <b>MCDF</b> )	-
	Cn	<b>113</b>	<b>114</b>	<b>115</b>	<b>116</b>	<b>117</b>	<b>118</b>
IHFSCC (eV)	-	0.680 <sup>1</sup>	-	0.368	0.905	1.589	0.056 <sup>3</sup>
						1.45 <sup>2</sup> ( <b>MCDF</b> )	



	Fr	Ra
IP calc (eV)	0.491 <sup>4</sup>	<b>0.100</b>
IP exp. (eV)	0.492	
	<b>119</b>	<b>120</b>
IP calc (eV)	0.663 <sup>4</sup>	<b>0.052</b>

<sup>1</sup> E. Eliav *et al.*, PRA **53**, 3926 (1996)

<sup>2</sup> Z. Chang *et al.*, JPC A **114**, 13388 (2010)

<sup>3</sup> I. Goidenko, PRA **67**, 020102 (2003)

<sup>4</sup> A. Landau *et al.* JCP **115**, 2389 (2001)

# Elements 121 and 122

	Ac	Th
Ground state	$7s^2 \mathbf{6d}_{3/2}$	$\mathbf{6d}^2 7s^2$
IP (eV)	5.17 ( $7s^2$ )	6.307 ( $\mathbf{6d}^2 \mathbf{7s}$ )
EA (eV)	0.345 ( $7s^2 \mathbf{6d} \mathbf{7p}$ )	-
	<b>121</b>	<b>122</b>
Ground state	$8s^2 \mathbf{8p}_{1/2}$	$8s^2 \mathbf{7d}_{3/2} \mathbf{8p}_{1/2}$
IP (eV)	4.447 ( $8s^2$ )	5.595 ( $8s^2 \mathbf{7d}_{3/2}$ )
EA (eV)	0.569 ( $8s^2 \mathbf{7p}^2_{1/2}$ )	-

<sup>1</sup> E. Eliav, S. Shmulyian, U. Kaldor, and Y. Ishikawa, J. Chem. Phys. **109**, 3954 (1998)

<sup>2</sup> E. Eliav, A. Landau, Y. Ishikawa, and U. Kaldor, J. Phys. B **35**, 1693 (2002)

# Polarizabilities

## Finite field approach

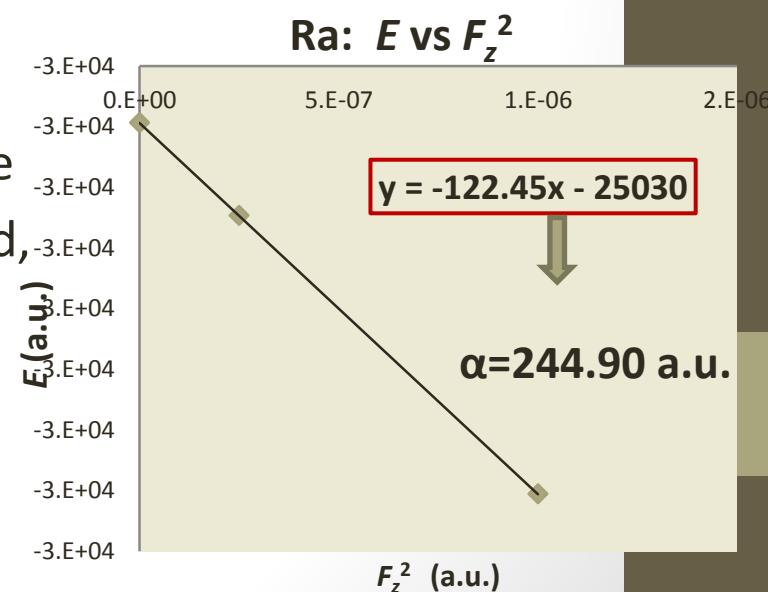
- Electric dipole polarizability ( $\alpha$ ): a measure of the distortion of the charge distribution by an external electric field. For a uniform electric field acting in the  $z$  direction ( $F_z$ ),

$$\alpha = -\frac{\partial^2 E(F_z)}{\partial F_z}$$

- For an atom in an external electric field,

$$E(F_z) = E(0) + F_z \left. \frac{\partial E(F_z)}{\partial F_z} \right|_{F_z=0} + \frac{1}{2} F_z^2 \left. \frac{\partial E^2(F_z)}{\partial^2 F_z} \right|_{F_z=0} + \dots$$

- Calculate the energy  $E(F_z)$  of the ground state of the atom for different values of external field, ( $F_z=0.0, 0.0005, 0.001$  a.u.)
- Obtain derivative with respect to  $F_z^2$  ( $\alpha$ ) by numerical differentiation
- Polarizability:  $\alpha=-2 \cdot a$



# Polarizabilities – 7p and 8s elements

	Hg	Tl	Pb	Bi	Po	At	Rn
This work (a.u.)	34.15	52.3	46.96	47.36	<b>43.57</b>	<b>42.37</b>	35.04
$\alpha$ other calc. (a.u.)	34.42 <sup>1</sup> <b>PP+CCSD(T)</b>	51.6 <sup>2</sup> <b>4C+CI</b>	47.71 <sup>3</sup> <b>RCCSD(T)</b>	48.6 <sup>4</sup> <b>DK+CASPT2</b>	46.8 <sup>5</sup> <b>DF</b>	43.14 <sup>6</sup> <b>2C+CI</b>	28.61 <sup>7</sup>
$\alpha$ exp. (a.u.)	33.9 (0.3)	51(7)	47.1 (7)				
	<b>Cn</b>	<b>113</b>	<b>114</b>	<b>115</b>	<b>116</b>	<b>117</b>	<b>118</b>
This work (a.u.)	27.64	29.9	30.6	<b>73.69</b>	<b>72.00</b>	<b>71.43</b>	46.3
$\alpha$ other calc.	25.8 <sup>1</sup> <b>PP+CCSD(T)</b>		31.98 <sup>3</sup> <b>RCCSD(T)</b>				52.4 <sup>7</sup> <b>ECP-CCSD(T)</b>

<sup>1</sup>M. Seth *et al.*, JCP **106**, 3623 (1997)

<sup>2</sup>T. Fleig, PRA **72**, 0526506 (2005)

<sup>3</sup>Thierfelder *et al.*, PRA **78**, 052506 (2008)

<sup>4</sup>B. O. Roos *et al*, JPC **108**, 2851 (2004)

<sup>5</sup>V. Kellö and A. J. Sadlej, Theor. Chim. Acta **83**, 351 (1992)

<sup>6</sup>T. Fleig and A. J. Sadlej, PRA **65**, 032506 (2002)

<sup>7</sup>C. Nash, JPC A **109**, 3493 (2005)

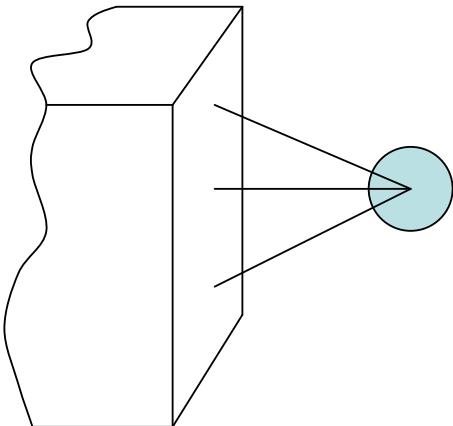
<sup>8</sup>I. Lim *et al.*, JCP **122**, 104103 (2005)

<sup>9</sup>I. Lim and P. Schwerdtfeger, PRA **70** 062501 (2004)

	Fr	Ra
$\alpha$ this work [a.u.]	<b>in progress</b>	244.90
$\alpha$ <b>PP/DK+CCSD(T)</b>	315.2 <sup>8</sup>	246.2 <sup>9</sup>
	<b>119</b>	<b>120</b>
$\alpha$ this work [a.u.]	<b>in progress</b>	168.15
$\alpha$ <b>DK+CCSD(T)</b>	163.8 <sup>8</sup>	

# Adsorption on inert surfaces:

Weak interactions: ad-atom - slab model



$$E(x) = -\frac{3}{16} \left( \frac{\varepsilon - 1}{\varepsilon + 2} \right) \frac{\alpha_{at}}{\left( \frac{1}{IP_{slab}} + \frac{1}{IP_{at}} \right)} x^3 \quad \square -\Delta H_{ads}$$

$E(x)$ -dispersion interaction energy

$IP_{slab}$  - ionization potentials of the surface material

$\varepsilon$  - the dielectric constant of the surface material

$IP_{at}$  - ionization potentials of the adsorbed atom

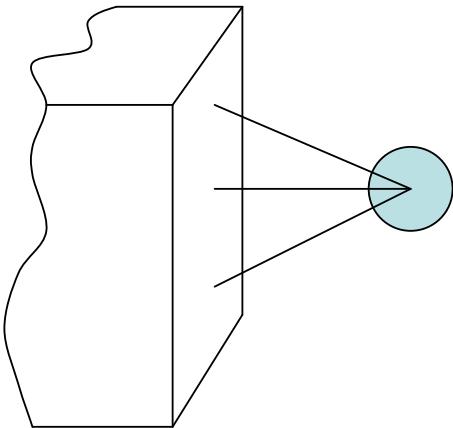
$\alpha_{at}$  - polarizability of the adsorbed atom

$x$  - the adatom-surface distance (approximated by  $R_{vdW}$  of the adsorbed species)

} Exp. values  
} Calc. values

# Adsorption on inert surfaces:

Weak interactions: ad-atom - slab model



$$E(x) = -\frac{3}{16} \left( \frac{\varepsilon - 1}{\varepsilon + 2} \right) \frac{\alpha_{at}}{\left( \frac{1}{IP_{slab}} + \frac{1}{IP_{at}} \right)} x^3 \quad \square -\Delta H_{ads}$$

$E(x)$ -dispersion interaction energy

$IP_{slab}$  - ionization potentials of the surface material

$\varepsilon$  - the dielectric constant of the surface material

$IP_{at}$  - ionization potentials of the adsorbed atom

$\alpha_{at}$  - polarizability of the adsorbed atom

$x$  - the adatom-surface distance (approximated by

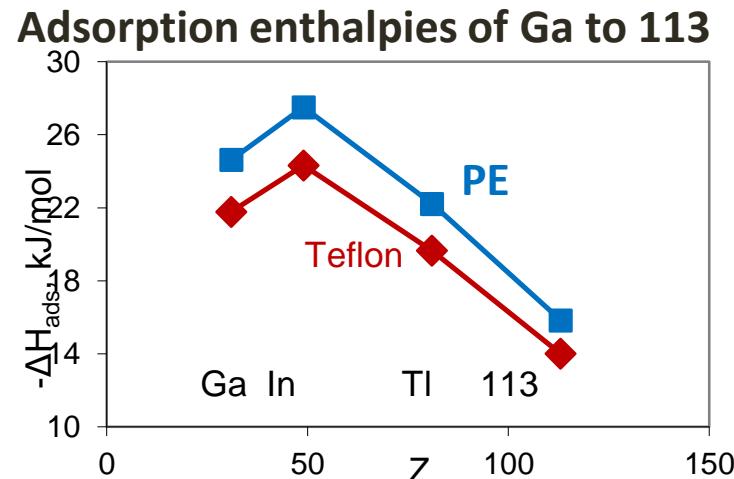
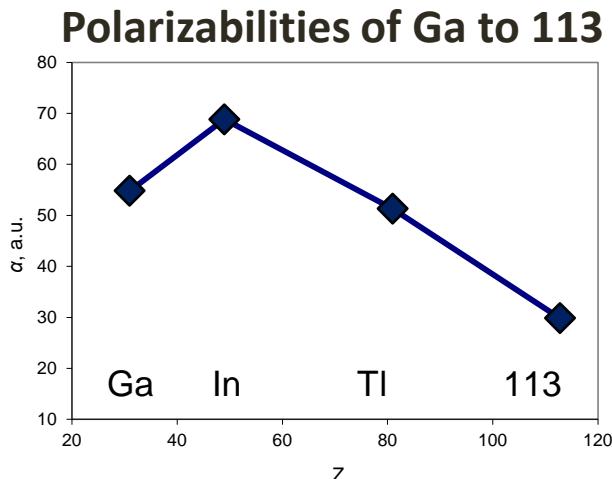
$R_{vdW}$  of the adsorbed species)

} Exp. values  
} Calc. values

# Example: adsorption of Tl and 113 on Teflon and PE

	$\alpha$ [a.u.]	IP [eV]	$R_{vdW}$ [Å]	$\Delta H_{ads}$ (Teflon) [kJ/mol]	$\Delta H_{ads}$ (PE) [kJ/mol]
Al	45.89	5.986	1.89	17.28	19.54
Ga	54.80	5.999	1.87	21.78	24.63
In	68.83	5.786	1.93	24.32	27.51
Tl	51.3	6.108	1.90	19.65	22.22
113	29.85	7.306	1.84	14.00	15.83

	Teflon	PE
$\epsilon$	2.04	2.26
IP <sub>slab</sub> [eV]	10.12	9.7

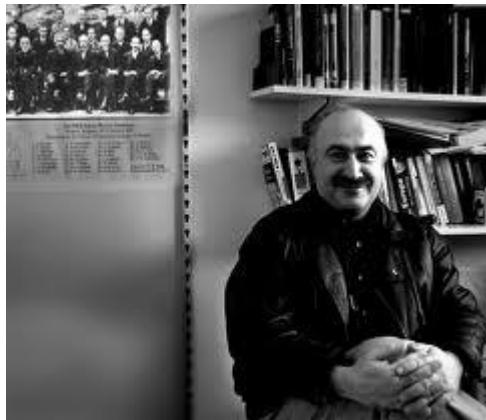


# Summary

- Methods:
  - 4-c Dirac-Coulomb-Breit Hamiltonian
  - Fock space CC
  - Intermediate Hamiltonian
  - Large, converged basis sets
- Powerful computational tool kit, tailored for superheavy systems
- Benchmark calculations for elements 112 to 122, and lighter homologues
- Excellent agreement with available experiment; same accuracy is expected for SHE

# Collaboration:

- A. Borshevsky – Centre for Theoretical Chemistry and Physics, Massey University, Auckland, New Zealand (at the moment visiting researcher at HIM)
- Valeria Pershina- GSI, Darmstadt, Germany
- Ephraim Eliav and Uzi Kaldor – Tel Aviv University, Israel



**Many thanks to the Helmholtz-Institute Mainz  
for the visiting scientist grant**

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- A. Borschevsky – Centre for Theoretical Chemistry and Physics, Massey University, Auckland, New Zealand (at the moment visiting re
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**Thank you for your  
attention!**



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